

NEOLIGNANS FROM *VIROLA CARINATA* FRUIT*

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(Revised received 21 August 1984)

Key Word Index—*Virola carinata*, Myristicaceae, fruits, benzodioxan neolignans, eusiderin-E, β -aryloxy-arylpropane neolignans

Abstract—The fruits of *Virola carinata* contain the lignans (–)-cubebin, (–)-hinokinin and (+)-asarinin, besides four neolignans of the β -aryloxy-arylpropane type and two neolignans of the benzodioxane (eusiderin) type

INTRODUCTION

Virola carinata is one of the most widespread and abundant Myristicaceae species in the low regions of the Amazonian valley. Several of its parts have indigenous medicinal uses [2]. Previous studies demonstrated the occurrence in the trunk wood of 7,4'-dimethoxyflavone besides three tetralin type neolignans (+)-guaiacin (1a), (–)-isootobaphenol (1b) and (–)-galcatin (1c) [3], and in the bark of tryptamines [2] besides two biphenyl type neolignans, dehydrodieugenol (2a) and *O*-methyl-dehydrodieugenol (2b) [4], of three β -aryloxy-arylpropane type neolignans, (–)-carinatone (3a), carinatol (3b) and carinatol (3c), of four benzofuran type neolignans carinatine (4a), carinatidin (4b), dihydrocarinatol (4c) and dihydrocarinatidin (4d), of a β -aryloxy-arylpropanol type neolignan carinatidiol (5) and of a diaryl ether type neolignan dehydrodieugenol-B (6) [5–7]. The present study revealed the presence in the fruit kernel of three known lignans, (–)-cubebin (7a), (–)-hinokinin (7b) and (+)-asarinin (8), all isolated previously from *V. multinervia* [8], besides the β -aryloxy-arylpropane type neolignan 9a and the benzodioxane type neolignan eusiderin, here designated eusiderin-A (10a). The latter compound, isolated previously from Myristicaceae [9] and Lauraceae [10] species, together with additional β -aryloxy-arylpropane type neolignans (11a, 11b, 11d) and another benzodioxane type neolignan (12) was also located in the pericarp.

In order to facilitate comparisons among the different compounds numbering of neolignans follow the biogenetic rules outlined in a review [10].

RESULTS

Spectral comparisons (Tables 1 and 2) classified compound 12, $C_{18}H_{14}O_2 \cdot OH(OMe)_3$, into the group of

benzodioxane neolignans. Its major distinction from the known eusiderins A (10a), B (10b) [10], C (10c) and D (10d) [11] refers to the presence of a propenyl instead of an allyl side chain. The aryl/Me substituents are *cis*-oriented in eusiderins C and D and *trans*-oriented in eusiderins A and B, as in the novel compound 12 now designated eusiderin E. Besides, the ORD curves of eusiderins A and E show positive Cotton effects around 300 nm.

Spectral comparisons (Tables 3–6) classified compounds 9a, 11a, 11b and 11d into the group of 8-O-4'-neolignans. Compounds 9a, $C_{18}H_{15}O(OMe)_3CH_2O_2$, 11a, $C_{18}H_{15}O(OMe)_4OH$, and the model compound 9b, $C_{18}H_{15}O(OMe)_5$, a constituent of *Myristica fragrans* [12], share all 1H NMR features due to the basic neolignan skeleton (Table 3), including the benzylic methylene bands, with the exception of the allyl derived signals in 9a and 9b vs the propenyl derived signals in 11a.

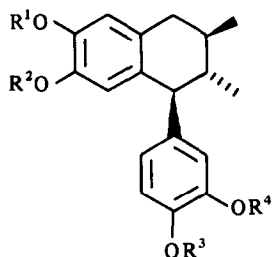
Compound 11b, $C_{18}H_{14}O(OMe)_4(OH)_2$, and the model compounds 11c, $C_{18}H_{15}O(OMe)_4OH$, a constituent of *V. surinamensis* [13], and 9c, $C_{18}H_{14}O(OMe)_5OH$, a constituent of *M. fragrans* [12], share many 1H NMR features due to the basic neolignan skeleton (Table 4), including the benzylic oxymethine bands. The H-7, H-8 coupling constants for the *threo*-derivative 11c ($J = 8$ Hz [13]) and the *erythro*-derivative 9c ($J = 3$ Hz [12]), however, are diagnostically different and the novel 11b ($J = 4$ Hz) must thus exist in the *erythro*-configuration. Furthermore 11b and 11c are characterized by propenyl substituents, while 9c is characterized by an allyl substituent.

Compound 11d, $C_{18}H_{13}O_2(OMe)_4OH$, and the synthetic model compounds 11e, $C_{18}H_{14}O_2(OMe)_4$ [13], and 9d, $C_{18}H_{13}O_2(OMe)_5$ [12], share many features, including the conjugated carbonyl bands ($IR \nu_{max}$ 1680 cm^{-1}) and the α -carbonylic oxymethine bands (Table 5). Here 11d and 11e possess propenyl substitution and 9d allyl substitution.

Comparisons with the model compounds show furthermore that all four new compounds, 9a, 11a, 11b, 11d, have identical oxygenation patterns and that, as expected, the NMR signals of H-2 and H-6 are shifted by ca 0.9 ppm to lower field (δ 7.5) upon oxidation of the benzylic position to a ketone. Thus the three OMe groups and the methylenedioxy-group of 9a can only be located as shown

*Part XXV in the series "The Chemistry of Brazilian Myristicaceae". For Part XXIV see ref. [1]. Taken from part of the doctorate thesis presented by S. H. C. to Universidade de São Paulo (1983).

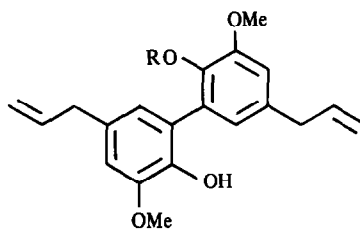
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1a $R^1 = R^4 = \text{Me}, R^2 = R^3 = \text{H}$

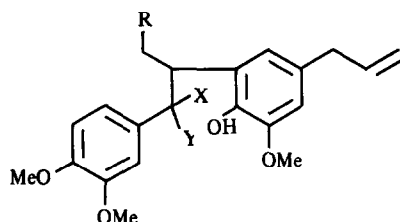
1b $R^1 - R^2 = \text{CH}_2, R^3 = \text{H}, R^4 = \text{Me}$

1c $R^1 - R^2 = \text{CH}_2, R^3 = R^4 = \text{Me}$



2a $R = \text{H}$

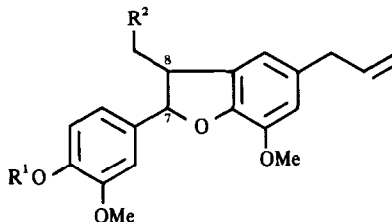
2b $R = \text{Me}$



3a $R = \text{H}, X - Y = \text{O}$

3b $R = \text{OH}, X - Y = \text{O}$

3c $R = X = \text{H}, Y = \text{OH}$

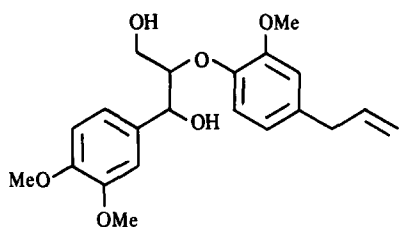


4a $R^1 = \text{Me}, R^2 = \text{H}, \Delta^{7,8}$

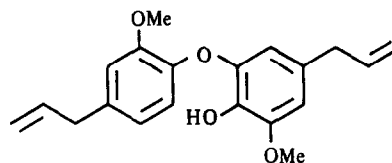
4b $R^1 = R^2 = \text{H}, \Delta^{7,8}$

4c $R = \text{Me}, R^2 = \text{OH}$

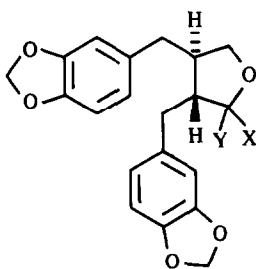
4d $R^1 = R^2 = \text{H}$



5

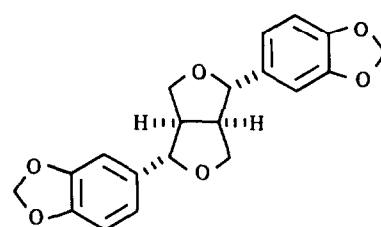


6



7a $X = \text{H}, Y = \text{OH}$

7b $X - Y = \text{O}$



8

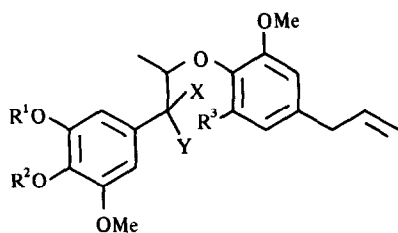
in the formula. The ^{13}C NMR spectra (Table 6) demonstrate that the pairs of aromatic rings in **11a**, **11b** and **11d** are symmetrically substituted, a fact which defines the substitution of each pair by four OMe groups and one OH group as shown in the formulae.

In accordance with these deductions, all mass spectra show intense peaks at m/z 194 and 193 assigned to ions derived from the oxydimethoxy-allyl (**9a**) and propenyl (**11a**, **11b**, **11d**) parts of the molecules. An additional

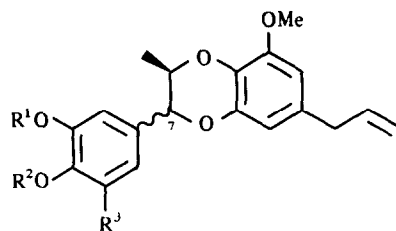
strong peak appears in this region (m/z 192) only in the case of the allyl-derivative (**9a** → **13**). Peaks at m/z 165 and 167 are assigned to tropylium ions *ex* **9a** and **11a**, while peaks at m/z 181 are caused by arylacylium ions *ex* **11b** and **11d**.

DISCUSSION

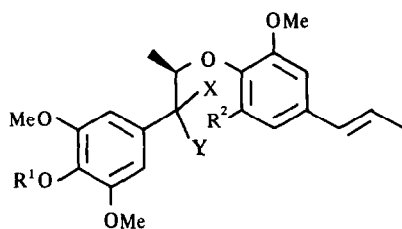
The data obtained so far indicate a differential distribution to exist for lignoids of different parts of *V. carinata*.



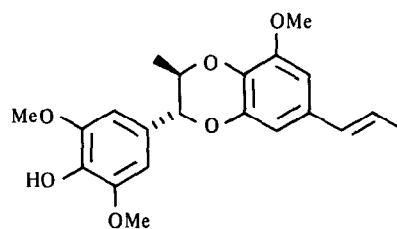
- 9a** $R^1 - R^2 = \text{CH}_2$ $R^3 = \text{OMe}$, $X = Y = \text{H}$
9b $R^1 = R^2 = \text{Me}$ $R^3 = \text{OMe}$ $X = Y = \text{H}$
9c $R^1 = R^2 = \text{Me}$ $R^3 = \text{OMe}$, $X = \text{H}$, $Y = \text{OH}$ (*erythro*)
9d $R^1 = R^2 = \text{Me}$ $R^3 = \text{OMe}$, $X - Y = \text{O}$



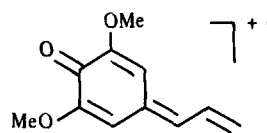
- 10a** $R^1 = R^2 = \text{Me}$ $R^3 = \text{OMe}$, $7 - \alpha\text{Ar}$
10b $R^1 - R^2 = \text{CH}_2$, $R^3 = \text{H}$, $7 - \alpha\text{Ar}$
10c $R^1 = R^2 = \text{Me}$, $R^3 = \text{OMe}$, $7 - \beta\text{Ar}$
10d $R^1 = R^2 = \text{Me}$, $R^3 = \text{H}$ $7 - \beta\text{Ar}$



- 11a** $R^1 = X = Y = \text{H}$, $R^2 = \text{OMe}$
11b $R^1 = X = \text{H}$, $R^2 = \text{OMe}$, $Y = \text{OH}$ (*erythro*)
11c $R^1 = \text{Me}$, $R^2 = Y = \text{H}$, $X = \text{OH}$ (*threo*)
11d $R^1 = \text{H}$ $R^2 = \text{OMe}$ $X - Y = \text{O}$
11e $R^1 = \text{Me}$ $R^2 = \text{H}$, $X - Y = \text{O}$



12



13

Table 1 ^1H NMR chemical shifts (δ) of benzodioxane neolignans **10a**, **10b**, **12** (60 MHz, CDCl_3), **10c** and **10d** (270 MHz, CDCl_3)

H	10a, b [11]		10c, d [9]		12	
	δ	J	δ	J	δ	J
9	1.30 d	6	1.15 d	6.5	1.25 d	6
8	3.9-4		4.60 dq	2.5, 6.5	4.05-4.25	
7	4.58 d	8	5.1 d	2.5	4.55 d	8
9'	4.9-5.3		5.1-5.2		1.85 d	6
8'	5.6-6.3		5.98 ddt	6.5, 10.6, 17	5.8-6.2	
7'	3.31 d	7	3.31 d	7	6.45-6.75	

Coupling constants (J) in Hz

Trunk wood and bark contain derivatives of dioxy-allyl and propenylphenols. The compounds from wood belong to the 2,7,8,8'-neolignan type (1), while the compounds from bark belong to the 3,3'- (2), 8,3'- (3), 7-O-4',8,3'- (4),

Table 2 ^{13}C NMR chemical shifts (δ) of benzodioxane neolignans (20 MHz, CDCl_3)

C	10a, b [11] δ	10c, d [9] δ	12 δ
9	172 $\pm$ 0.1	12.6	17.3
8	74.1	73.2	74.3
7	81.4 $\pm$ 0.2	77.1	81.1
9'	115.7 $\pm$ 0.1	115.9	18.2
8'	137.3 $\pm$ 0.1	137.5	124.3
7'	40.0	40.0	130.7

8-O-4'- (5) and 3-O-4'- (6) neolignan types. The fruit-kernels contain lignans of identical oxygenation pattern (7, 8). All other constituents of the fruit, however, are derivatives of trioxy-allyl and propenylphenols belonging to 8-O-4'- (9, 11) and 7-O-3',8-O-4'- (10, 12) neolignan types. Kernels seem to accumulate preferentially neolignans formed by the oxidative coupling of propenyl- and

Table 3 ^1H NMR chemical shifts (δ) of β -aryloxy-arylpropane neolignans (80 MHz, CDCl_3)

H	9b [12]		9a		11a	
	δ	J	δ	J	δ	J
9	1 23 <i>d</i>		1 25 <i>d</i>	6	1 25 <i>d</i>	6
8	3 82 <i>m</i>		4 1–4 5		4 1–4 5	
7	3 10 <i>dd</i>	5 5, 7 7	3 05 <i>dd</i>	6, 14	3 10 <i>dd</i>	6, 14
			2 70 <i>dd</i>	8, 14	2 70 <i>dd</i>	8, 14
9'	5 06 <i>m</i>		4 9–5 2		1 85 <i>d</i>	
8'	5 90 <i>m</i>		6 6–6 25		6 1–6 3	
7'	3 33 <i>brd</i>		3 30 <i>d</i>	7	6 5–6 75	

Coupling constants (J) in HzTable 4 ^1H NMR chemical shifts (δ) of *erythro* (9c, 11b) and *threo* (11c) β -aryloxy- α -hydroxy-arylpropane neolignans (60, 80, 100 MHz, respectively, in CDCl_3)

H	9c [12]		11b		11c [13]	
	δ	J	δ	J	δ	J
9	1 13 <i>d</i>		1 15 <i>d</i>	6	1 19 <i>d</i>	6
8	4 35 <i>dq</i>		4 3–4 5		4 14 <i>m</i>	
7	4 80 <i>d</i>	3	4 75 <i>d</i>	4	4 62 <i>d</i>	8
9'	5 11 <i>m</i>		1 85 <i>d</i>	6	1 87 <i>d</i>	5 5
8'	5 99 <i>m</i>		6 1–6 4		6–6 3	
7'	3 38 <i>brd</i>		6 8–6 9		6 37 <i>d</i>	16

Coupling constants (J) in HzTable 5 ^1H NMR chemical shifts (δ) of β -aryloxy- α -oxo-arylpropane neolignans 9d, 11e, 11d (resp 60, 100, 80 MHz, CDCl_3)

H	9d [12]		11e [13]		11d	
	δ	J	δ	J	δ	J
9	1 57 <i>d</i>	6 8	1 71 <i>d</i>	7	1 55 <i>d</i>	7
8	5 30 <i>q</i>	6 8	5 33 <i>q</i>	7	5 25 <i>q</i>	7
9'	5 06 <i>m</i>		1 83 <i>d</i>	6	1 87 <i>d</i>	6
8'	5 90 <i>m</i>		6–6 4		5 8–6 3	
7'	3 30 <i>brd</i>				6 5–6 8	

Coupling constants (J) in Hz

allylphenols (9a, 10a), while in the pericarp one sole such derivative (10a) is accompanied by four neolignans formed by the oxidative coupling of two propenyl derivatives (11a, 11b, 11d, 12)

EXPERIMENTAL

Isolation of constituents Unripe fruits of *V. carinata* (Benth) Warb were collected by Hipólito F Paulino Fo on Feb 2, 1981, near km 126 of the Santarém-Cuiabá road, Municipality of Santarém, Pará State. The fruits were separated into kernels,

Table 6 ^{13}C NMR chemical shifts (δ) of β -aryloxy-arylpropane neolignans (20 MHz, CDCl_3)

C	9a	11a	11b	11d	11c [13]
9	19 6	19 6	12 8	17 6	17 0
8	79 5	79 9	73 3	80 2	78 5
7	43 3	43 4	82 6	196 5	83 6
9'	115 8	18 2	18 2	17 9	18 2
8'	137 4	124 9	125 5	124 9	124 6
7'	40 5	130 0	130 0	130 5	130 2
1	133 5	131 1	131 1	126 3	137 6
2	109 2	103 3	103 1	106 8	104 2
3	148 7	147 0	147 1	146 3	152 9
4	135 4	133 3	134 1	139 6	135 5
5	143 4	147 0	147 1	146 3	152 9
6	105 8	103 3	103 1	106 8	104 2
1'	133 6	131 1	134 0	134 4	133 3
2'	103 3	106 5	104 5	102 6	109 1
3'	153 9	153 8	153 7	152 9	150 5
4'	134 5	133 5	134 1	133 6	146 4
5'	153 7	153 8	153 7	152 9	118 6
6'	103 7	106 5	104 5	102 6	118 7
MeO-3	—	56 0	56 1	55 0	56 0
MeO-5	56 6	—	—	—	—
MeO-4	—	—	—	—	60 6
MeO-3'	—	56 0	56 3	56 3	55 6
MeO-5'	—	—	—	—	—
CH_2O_2 -3,4	101 1	—	—	—	—

pericarps, arils and teguments. Kernels (150 g) and pericarps (75 g) were percolated with CHCl_3 . The extracts (86 g and 38 g, respectively) were crystallized from MeOH to glycerides (71 g and 28 g, respectively). The mother-liquors were evapd. The residue from the kernels (15 g) was partitioned between petrol and MeOH– H_2O (9/1). The petrol soln was evapd and the residue (11.5 g) was submitted to CC (150 g silica gel). Elution with petrol–EtOAc (9/1) gave in order six fractions. Fr 1 (3.5 g) was composed of aliphatic material. Frs 2 (1.1 g) and 3 (3.2 g) were crystallized from C_6H_{14} – CHCl_3 to glycerides (0.4 g and 1.7 g, respectively). The mother-liquors were evapd. The residue of Fr 2 (0.6 g) was separated by prep TLC (silica gel, C_6H_6 –EtOAc 9/1) into 8 (14 mg) and 9a (14 mg). The residue of Fr 3 was composed of 10a (1.5 g). Fr 4 (2 g) and 5 (1.4 g) were separated by prep TLC (silica gel, C_6H_6 –EtOAc 9/1 and 19/1, respectively) into 7b (0.5 g) and 7a (0.1 g), respectively. The residue from the pericarps (10 g) was submitted to CC (100 g silica gel). Elution with C_6H_6 –EtOAc (4/1) gave in order five fractions. Fr 1 (3.1 g) was composed of aliphatic material. Frs 2 (1.5 g), 3 (0.8 g), 4 (1.4 g) and 5 (0.5 g) were separated by prep TLC (silica gel, C_6H_6 –EtOAc 9/1) into 10a (26 mg) and 11a (26 mg), 11a (88 mg) and 12 (48 mg), 11b (140 mg) and 11c (140 mg), 11b (100 mg) and 11c (30 mg), respectively.

rel-(7R,8R)- Δ^7 -4-Hydroxy-3,5,5'-trimethoxy-7-O-3',8-O-4'-neolignan (equisiderin-E, 12). Oil, UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm 210, 275 (ϵ 11 100, 2150). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} 3500, 1610, 1590, 1505, 1460, 1330, 1270, 1220, 1150, 1100, 1040, 990, 960, 915. ^1H NMR (δ) Table 1 and 6 6 (*brs*, H-2, H-6, H-2', H-6'), 3 85 (*s*, 3 MeO). ^{13}C NMR δ Table 2 and 132 2 (C-1), 104 5 (C-2), 147 4 (C-3), 135 6 (C-4), 147 4 (C-5), 104 5 (C-6), 132 4 (C-1'), 107 4 (C-2'), 144 6 (C-3'), 131 9 (C-4'), 148 7 (C-5'), 102 2 (C-6'), 56 6 (MeO-3, MeO-5), 56 2 (MeO-5'). MS m/z (rel int) 372 ($[\text{M}]^+$, 20), 256 (30), 194 (46), 193 (7), 192 (2), 191 (8), 181 (3), 165 (3), 164 (2), 163 (3), 135 (5), 74 (100).

ORD (MeOH) $[\phi]_{318}^{pk} - 5100$, $[\phi]_{282}^{tr} - 9050$, $[\phi]_{230}^{pk} - 4650$ (for ORD of eusiderin-A (10a) see ref [9])

Δ^8 -3,4-Methylenedioxy-5,3',5'-trimethoxy-8-O-4'-neolignan (9a) Oil, UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm 240, 267 (ϵ 16 850, 12 700) IR ν_{CHCl_3} cm^{-1} 1630, 1590, 1500, 1450, 1430, 1240, 1225, 1190, 1120, 1090, 1040, 920 ^1H NMR δ Table 3 and 6 50 (*br s*, H-2), 6 35 (*br s*, H-6, H-2', H-6'), 3 85 (*s*, MeO-5), 3 80 (MeO-3', MeO-5'), 5 85 (*s*, CH_2O_2) ^{13}C NMR (Table 6) MS m/z (rel int) $^+$ 386 ($[\text{M}]^+$, 23), 221 (3), 194 (78), 193 (100), 192 (40), 165 (17), 135 (14), 107 (4) ORD (MeOH) $[\phi]_{318}^{tr} - 4550$, $[\phi]_{282}^{pk} - 1950$, $[\phi]_{230}^{tr} - 2400$, $[\phi]_{250}^{tr} 0$, $[\phi]_{242}^{pk} + 8950$, $[\phi]_{234}^{tr} + 5300$

Δ^7 -4-Hydroxy-3,5,3',5'-tetramethoxy-8-O-4'-neolignan (11a) Oil, UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm 233, 267 (ϵ 16 500, 14 850) IR ν_{CHCl_3} cm^{-1} 3500, 1610, 1580, 1500, 1460, 1330, 1270, 1240, 1220, 1140, 1120, 1040, 960, 920 ^1H NMR δ Table 3 and 6 45 (*s*, H-2, H-6), 6 55 (*s*, H-2', H-6'), 3 80 (*s*, MeO-3, MeO-5), 3 85 (*s*, MeO-3', MeO-5') ^{13}C NMR (Table 6) MS m/z (rel int) $^+$ 388 (M^+ , 8), 372 (2), 221 (5), 195 (61), 194 (70), 193 (8), 192 (3), 191 (7), 167 (17), 165 (8), 164 (7), 163 (14), 135 (15) ORD (MeOH) $[\phi]_{350}^{pk} - 750$, $[\phi]_{272}^{tr} - 6300$, $[\phi]_{263}^{tr} 0$, $[\phi]_{258}^{pk} + 500$, $[\phi]_{254}^{tr} + 250$, $[\phi]_{236}^{pk} + 7000$, $[\phi]_{230}^{tr} 0$, $[\phi]_{220}^{tr} - 8950$

rel-(7S,8R)- Δ^7 -4,7-Dihydroxy-3,5,3',5'-tetramethoxy-8-O-4'-neolignan (11b) Oil, UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm 225, 267 (ϵ 15 150, 9700) IR ν_{CHCl_3} cm^{-1} 3500, 1610, 1580, 1500, 1460, 1330, 1270, 1240, 1220, 1140, 1120, 1050, 960, 900 ^1H NMR δ Table 4 and 6 60 (*s*, H-2, H-6), 6 55 (*s*, H-2', H-6'), 3 80 (*s*, MeO-3, MeO-5), 3 85 (*s*, MeO-3', MeO-5') ^{13}C NMR (Table 6) MS m/z (rel int) $^+$ 404 ($[\text{M}]^+$, 10), 402 (5), 386 (3), 211 (4), 194 (20), 193 (60), 192 (8), 181 (15), 165 (4), 164 (10), 163 (38), 153 (10), 135 (8) ORD (MeOH) $[\phi]_{350}^{pk} - 100$, $[\phi]_{286}^{tr} - 5300$, $[\phi]_{256}^{pk} - 250$, $[\phi]_{236}^{tr} - 750$, $[\phi]_{220}^{tr} 0$

Δ^7 -4-Hydroxy-3,5,3',5'-tetramethoxy-7-oxo-8-O-4'-neolignan (11d) Oil, UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm 225, 273, 295 (ϵ 15 450, 7300, 6750) IR ν_{CHCl_3} cm^{-1} 3400, 1680, 1590, 1520, 1510, 1460, 1420, 1330, 1275, 1240, 1220, 1150, 1125, 1050, 960, 910 ^1H NMR δ Table 5 and 7 55 (*s*, H-2, H-6), 6 50 (*br s*, H-2', H-6'), 3 80 (*s*, MeO-3, MeO-5), 3 85 (*s*, MeO-3', MeO-5') ^{13}C NMR (Table 6) MS m/z (rel int) $^+$ 402 ($[\text{M}]^+$, 26), 372 (3), 221 (16), 209 (3), 208 (3), 194 (45), 193

(100), 181 (45), 165 (7) ORD (MeOH) $[\phi]_{350}^{tr} 0$, $[\phi]_{286}^{tr} - 5300$, $[\phi]_{256}^{pk} - 250$, $[\phi]_{236}^{tr} - 750$, $[\phi]_{220}^{tr} 0$

Acknowledgements—Financial aid by Conselho Nacional de Desenvolvimento Científico e Tecnológico (CNPq), Financiadora de Estudos e Projetos (FINEP) and Fundação de Amparo à Pesquisa do Estado de São Paulo (FAPESP), as well as graduate fellowships (to S H C) by CNPq and Coordenação para o Aperfeiçoamento de Pessoal de Nível Superior (CAPES), are gratefully acknowledged

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